Molecular characterization of oligomeric pyrolysis compounds of ethyl acrylate-butyl acrylate copolymer using thermal desorption/pyrolysis DART-MS Chikako Takei^{*1}, Kenichi Yoshizawa¹, Sayaka Nakamura², Hiroaki Sato², Hajime Ohtani³; ¹ BioChromato, Inc., ² National Institute of Advanced Industrial Science and Technology, ³ Nagoya Institute of Technology

Introduction

Industrial acrylic polymers are often synthesized with various comonomers to realize desired functions and physical properties. Since physical properties are affected by not only the average comonomer composition but also its distribution, molecular characterization of polymers is important from the viewpoint of controlling physical properties.

In this study, we investigated an ethyl acrylate (EA)-butyl acrylate (BA) copolymer by thermal desorption/pyrolysis (TDP) /DART-MS which can rapidly detect both thermally desorbed intact oligomers and oligomeric pyrolysis products of polymers. The observed results were compared with those by pyrolysis (Py-)GC/MS and MALDI-MS.

Experimental

Sample: EA-BA copolymer

- produced through random-copolymerization
- the feed monomer ratio of EA:BA = ca. 1:1



Analytical methods:



References

1) H.Sato, S.Nakamura, K.Teramoto, T.Sato; JASMS, 2014, 25(8), 1346-1355 2) S.Tsuge, H.Ohtani, and C.Watanabe; Pyrolysis-GC/MS Data Book of Synthetic Polymers, Elsevier (2011) p. 90-93

Results and Discussions

weight components.



Conclusion

polymers together with Py-GC/MS and MALDI-MS complementarily.

TDP/DART-MS can show evolution profiles of volatilized compounds and/or pyrolysis products during the heating process (Fig. 2). In the lower temperature (Fig. 3(a)), the observed compounds were mainly low molecular weight components.

- In the middle temperature (Fig. 3(b)), several series of thermal desorbed oligomers (up to m/z 1800) were observed.
- In the higher temperature (Fig. 3(c)), the observed compounds were mainly low molecular weight pyrolysis products of higher molecular

TDP/DART mass spectrum of the copolymer observed at around 350 °C (b) were informatively interpreted using the KMD¹ plot (Fig. 4). The plot showed a parallelogram-like dot distribution, which is the typical pattern of random copolymers. The plot revealed that the largest oligomer detected in this region was 8-mer composed of EA/BA = 4/4.



Fig.4 KMD plot of mass spectrum of Fig.3(b) with base unit EA ($C_5H_8O_2$)

Compositional distribution of 5-mer showed normal distribution with the maximum around EA/BA = 3/2 and 2/3 (Fig.5 and Table 1). This result suggest that this sample should be a random copolymer having a monomer composition estimated as EA/BA = 50/50 (Fig. 6).

> Table 1 Relative peak area among 5-mers Chemical Error Relative peak Comonomer ratio m/z EA/BA formula * (mDa) area 518.2980 C₂₅H₄₄NO₁₀ 2.1 2.9 5/0 4/1 16.1 546.3282 $C_{27}H_{48}NO_{10}$ -1.0 574.3595 C₂₉H₅₂NO₁₀ 3/2 31.5 -0.9 2/3 31.7 602.3910 C₃₁H₅₆NO₁₀ -1.1 14.8 630.4219 C₃₃H₆₀NO₁₀ 0.8 1/4 0/5 658.4501 C₃₅H₆₄NO₁₀ 2.3 3.0 [·] [M + NH₄]+



Fig.6 Curve fitting of relative peak area for 5-mer between observed and theoretical.

TDP/DART-MS can individually provide mass spectral information of volatile low molecular weight compounds and oligomeric pyrolysis products of higher polymeric compounds evolved during the heating process. The observed distributions suggest that the EA-BA copolymer should be a random copolymer having a monomer composition estimated as EA:BA = ca. 1:1. And, it was confirmed that these results were supported by Py-GC/MS and SEC-MALDI-MS. The distributions of oligomeric products up to 8-mer observed by TDP/DART-MS were more informative than those by Py-GC/MS. Thus, TDP/DART-MS can be a useful analytical technique to characterize





SEC-MALDI

parallelogram-like dot distribution, which is the typical pattern of random copolymers (Fig. 10).







Fitting factor = $\Sigma (RI_{n(mea.)} - RI_{n(cal.)})$ $0 \le n \le 3$



Fig.11 Curve fitting of relative peak intensity for 8-mer between observed and theoretical.